



US007378162B2

**(12) United States Patent**  
**Jeong et al.****(10) Patent No.: US 7,378,162 B2****(45) Date of Patent: \*May 27, 2008****(54) ORGANIC ELECTROLUMINESCENCE  
DEVICES USING RED PHOSPHORESCENCE  
COMPOUNDS**

WO WO 03/040256 A2 5/2003

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Lamansky, S., et al., "Synthesis and Characterization of Phosphorescent Cyclometalated Iridium Complexes," 2001 American Chemical Society, Inorg. Chem. 2001, 40, pp. 1704-1711, Mar. 26, 2001.

PCT Search Report.

\* cited by examiner

**(73) Assignee:** **LG Electronics Inc.**, Seoul (KR)*Primary Examiner*—Kamal A Saeed*Assistant Examiner*—Joseph R Kosack**(\*) Notice:** Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 0 days.**(74) Attorney, Agent, or Firm**—Ked & Associates, LLP

This patent is subject to a terminal disclaimer.

**(57) ABSTRACT****(21) Appl. No.:** **11/128,421**

Red phosphorescence compounds and organic electro-luminescence device using the same are disclosed. In an organic electroluminescence device including an anode, a hole injecting layer, a hole transport layer, a light emitting layer, an electron transport layer, an electron injecting layer, and a cathode serially deposited on one another, the organic electroluminescence device may use a compound as a dopant of the light emitting layer.

**(22) Filed:** **May 13, 2005****(65) Prior Publication Data**

US 2006/0202194 A1 Sep. 14, 2006

**(30) Foreign Application Priority Data**

Mar. 8, 2005 (KR) ..... 10-2005-0019182

**(51) Int. Cl.****H05B 33/14** (2006.01)**C09K 11/06** (2006.01)**C07D 215/00** (2006.01)**(52) U.S. Cl.** ..... **428/690**; 428/917; 257/40;  
313/504; 546/4; 546/10**(58) Field of Classification Search** ..... 313/504;  
546/4, 2, 10; 257/40; 428/690, 917  
See application file for complete search history.**(56) References Cited****U.S. PATENT DOCUMENTS**

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Formula 1

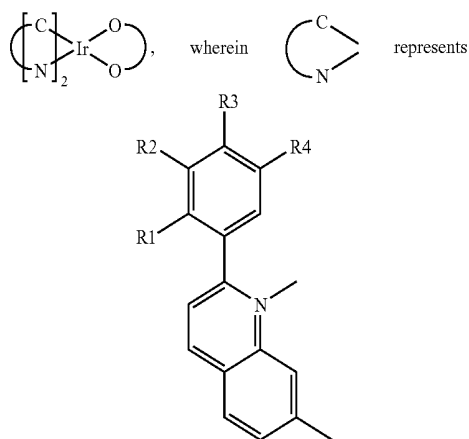
**7 Claims, 2 Drawing Sheets**

FIG. 1

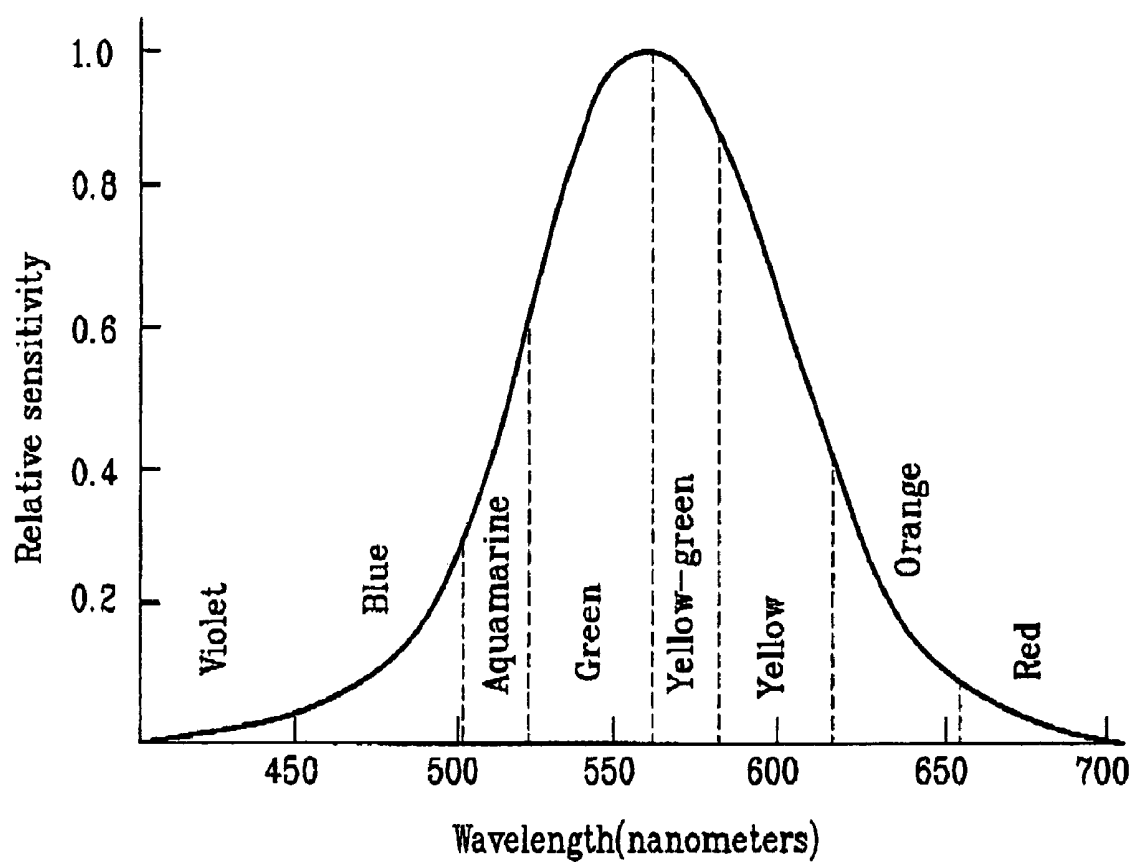
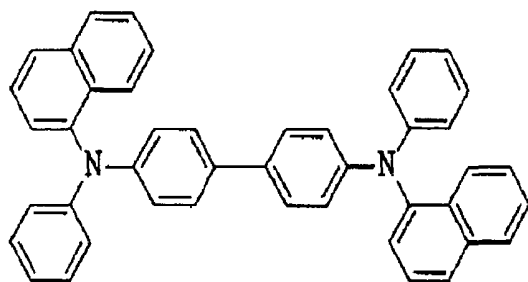
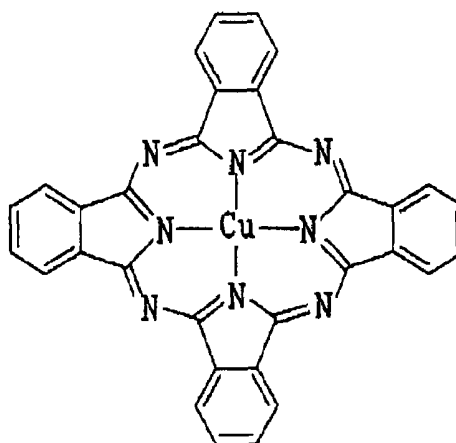


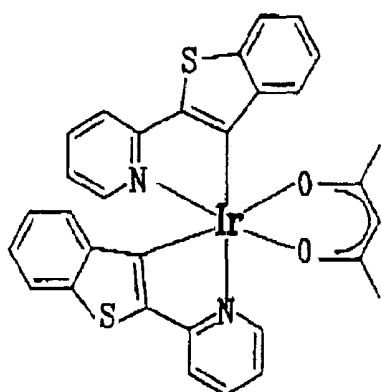
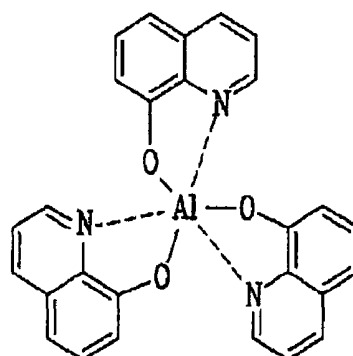
FIG. 2



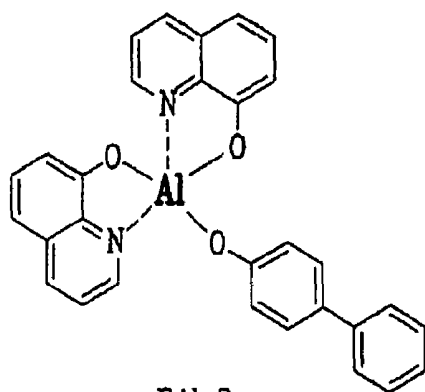
NPB



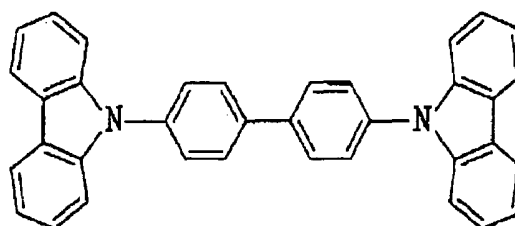
Copper(II) phthalocyanine (CuPc)

(btp)<sub>2</sub>Ir(acac)

Alq3



BAlq3



CBP

# ORGANIC ELECTROLUMINESCENCE DEVICES USING RED PHOSPHORESCENCE COMPOUNDS

This application claims the benefit of the Patent Korean 5  
Application No. 10-2005-0019182, filed on Mar. 8, 2005,  
which is hereby incorporated by reference as if fully set forth  
herein.

## BACKGROUND OF THE INVENTION

### 1. Field of the Invention

The present invention relates to an organic electrolumi-  
nescence device, and more particularly, to red phospho-  
rescence compounds and organic electroluminescence device 15  
using the same. Most particularly, the present invention  
relates to red phosphorescence being used as a dopant of a  
light emitting layer of an organic electroluminescence  
device, which is formed by serially depositing an anode, a  
hole injecting layer, a hole transport layer, a light emitting 20  
layer, an electron transport layer, an electron injecting layer,  
and a cathode.

### 2. Discussion of the Related Art

Recently, as the size of display devices is becoming larger,  
the request for flat display devices that occupy lesser space  
is becoming more in demand. Such flat display devices  
include organic electroluminescence devices, which are also  
referred to as an organic light emitting diode (OLED). 25  
Technology of such organic electroluminescence devices is  
being developed at a vast rate and various prototypes have  
already been disclosed.

The organic electroluminescence device emits light when  
electric charge is injected into an organic layer, which is  
formed between an electron injecting electrode (cathode)  
and a hole injecting electrode (anode). More specifically, 30  
light is emitted when an electron and a hole form a pair and  
the newly created electron-hole pair decays. The organic  
electroluminescence device can be formed on a flexible  
transparent substrate such as plastic. The organic electro-  
luminescence device can also be driven under a voltage 40  
lower than the voltage required in a plasma display panel or  
an inorganic electroluminescence (EL) display (i.e., a volt-  
age lower than or equal to 10V). The organic electrolumi-  
nescence device is advantageous in that it consumes less  
energy as compared to other display devices and that it 45  
provides excellent color representation. Moreover, since the  
organic EL device can reproduce pictures by using three  
colors (i.e., green, blue, and red), the organic EL device is  
widely acknowledged as a next generation color display  
device that can reproduce vivid images.

The process of fabricating such organic electrolumines-  
cence (EL) device will be described as follows:

- (1) An anode material is coated over a transparent sub-  
strate. Generally, indium tin oxide (ITO) is used as the  
anode material.
- (2) A hole injecting layer (HIL) is deposited on the anode  
material. The hole injecting layer is formed of a copper  
phthalocyanine (CuPc) layer having a thickness of 10  
nanometers (nm) to 30 nanometers (nm).
- (3) A hole transport layer (HTL) is then deposited. The 60  
hole transport layer is mostly formed of 4,4'-bis[N-(1-  
naphthyl)-N-phenylamino]-biphenyl (NPB), which is  
treated with vacuum evaporation and then coated to  
have a thickness of 30 nanometers (nm) to 60 nanom-  
eters (nm).
- (4) Thereafter, an organic light emitting layer is formed.  
At this point, a dopant may be added if required. In case

of green emission, the organic light emitting layer is  
generally formed of tris(8-hydroxy-quinolate)alumi-  
num (Alq<sub>3</sub>) which is vacuum evaporated to have a  
thickness of 30 nanometers (nm) to 60 nanometers  
(nm). And, MQD(N-Methylquinacridone) is used as  
the dopant (or impurity).

- (5) Either an electron transport layer (ETL) and an elec-  
tron injecting layer (EIL) are sequentially formed on  
the organic emitting layer, or an electron injecting/  
transport layer is formed on the organic light emitting  
layer. In case of green emission, the Alq<sub>3</sub> of step (4) has  
excellent electron transport ability. Therefore, the elec-  
tron injecting and transport layers are not necessarily  
required.

- (6) Finally, a layer cathode is coated, and a protective  
layer is coated over the entire structure.

A light emitting device emitting (or representing) the  
colors of blue, green, and red, respectively, is decided in  
accordance with the method of forming the light emitting  
layer in the above-described structure. As the light emitting  
material, an exciton is formed by a recombination of an  
electron and a hole, which are injected from each of the  
electrodes. A singlet exciton emits fluorescent light, and a  
triplet exciton emits phosphorescence light. The singlet exci-  
ton emitting fluorescent light has a 25% probability of  
formation, whereas the triplet exciton emitting phospho-  
rescence light has a 75% probability of formation. Therefore,  
the triplet exciton provides greater light emitting efficiency  
as compared to the singlet exciton. Among such phospho-  
rescence materials, red phosphorescence material may have  
greater light emitting efficiency than fluorescent materials.  
And so, the red phosphorescence material is being researched  
and studied as an important factor for enhancing the effi-  
ciency of the organic electroluminescence device.

When using such phosphorescence materials, high light  
emitting efficiency, high color purity, and extended durabil-  
ity are required. Most particularly, when using red phospho-  
rescence materials, the visibility decreases as the color purity  
increases (i.e., the X value of the CIE chromacity coordi-  
nates becomes larger), thereby causing difficulty in provid-  
ing high light emitting efficiency. Accordingly, red phos-  
phorescence material that can provide characteristics of  
excellent chromacity coordinates (CIE color purity of  
X=0.65 or more), enhanced light emitting efficiency, and  
extended durability needs to be developed.

## SUMMARY OF THE INVENTION

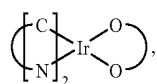
Accordingly, the present invention is directed to red  
phosphorescence compounds and an organic electro-lumi-  
nescence device using the same that substantially obviate  
one or more problems due to limitations and disadvantages  
of the related art.

An object of the present invention is to provide an organic  
electroluminescence device having high color purity, high  
brightness, and long durability by combining the compound  
shown in Formula 1, which is used as a dopant in a light  
emitting layer of the organic EL device.

Additional advantages, objects, and features of the inven-  
tion will be set forth in part in the description which follows  
and in part will become apparent to those having ordinary  
skill in the art upon examination of the following or may be  
learned from practice of the invention. The objectives and  
other advantages of the invention may be realized and  
attained by the structure particularly pointed out in the  
written description and claims hereof as well as the  
appended drawings.

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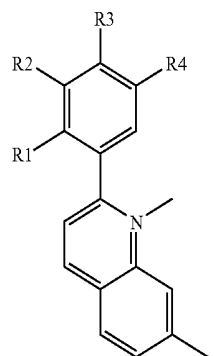
To achieve these objects and other advantages and in accordance with the purpose of the invention, as embodied and broadly described herein, a red phosphorescence compound is indicated as Formula 1 below:



wherein



represents

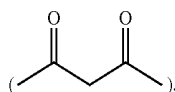


Herein each of R1, R2, R3, and R4 may be one of substituted or non-substituted C1 to C6 alkyl groups with disregard of one another. And, each of the C1 to C6 alkyl groups may be selected from a group consisting of methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, and t-butyl.

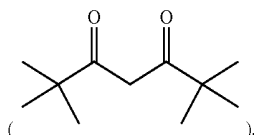
Additionally,



may include 2,4-pentanedione

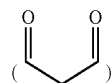


2,2,6,6,-tetra-methylheptane-3,5-dione



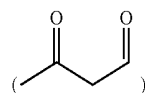
1,3-propanedione

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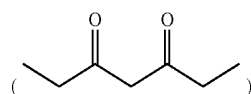


Formula 1

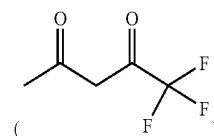
1,3-butanedione



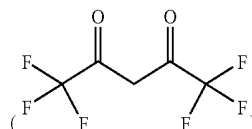
3,5-heptanedione



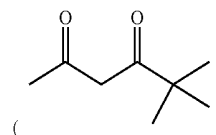
1,1,1-trifluoro-2,4-pentanedione



1,1,1,5,5,5-hexafluoro-2,4-pentanedione



and 2,2-dimethyl-3,5-hexanedione

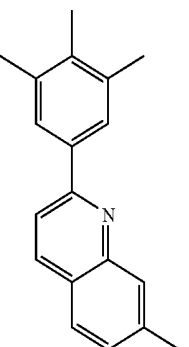
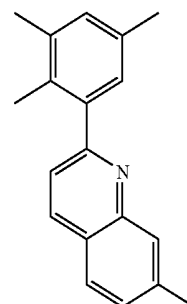
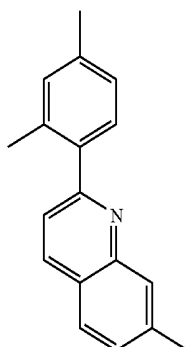
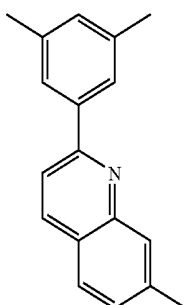
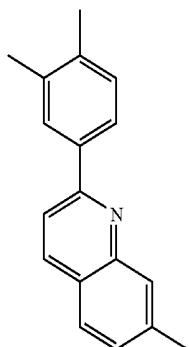
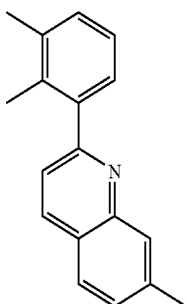
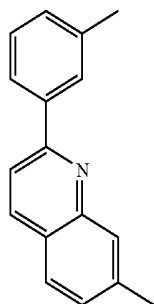
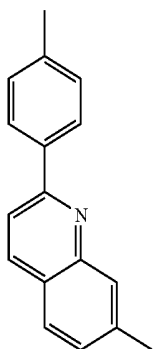
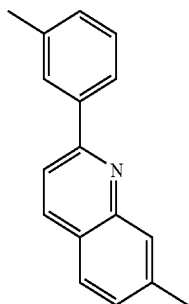
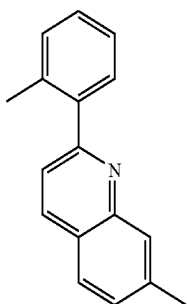


Moreover,



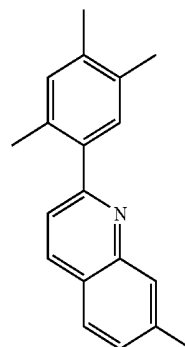
may be any one of the following chemical formulas:

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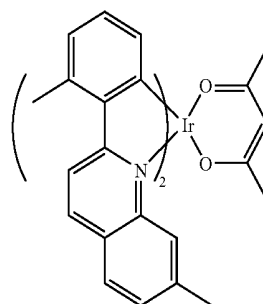


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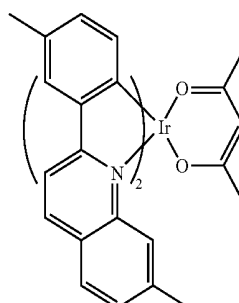
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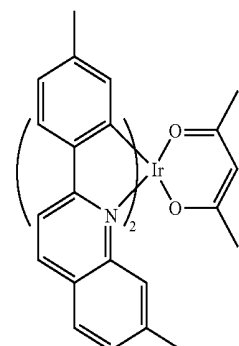
Furthermore, the Formula 1 may be any one of the following chemical formulas:



A-1



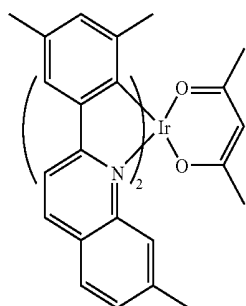
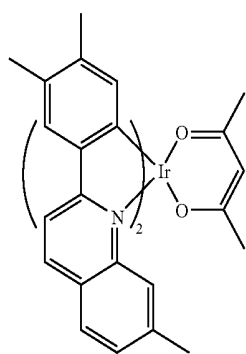
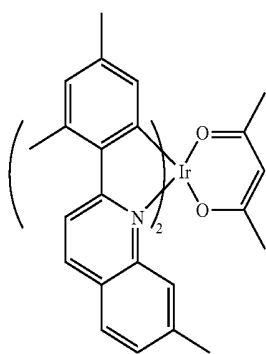
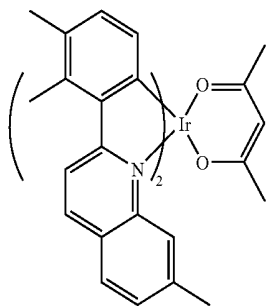
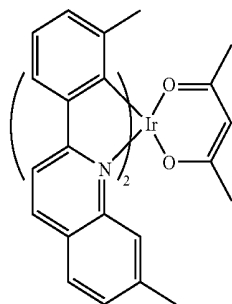
A-2



A-3

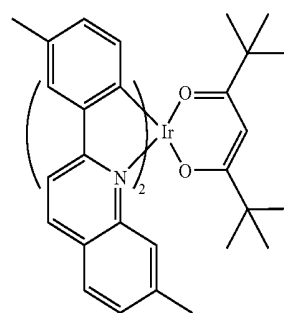
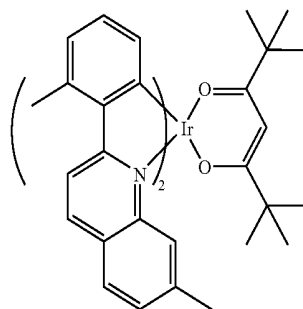
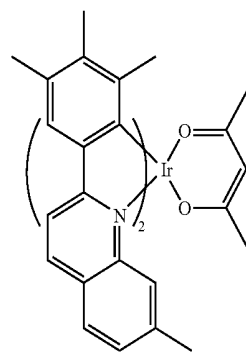
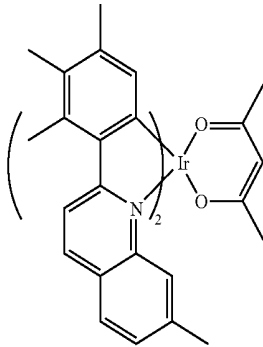
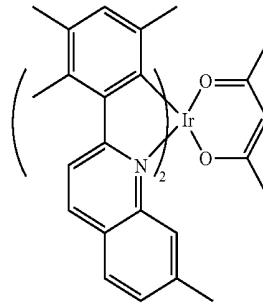
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A-4

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A-5 15

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A-6

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A-7 40

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A-8

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A-9

A-10

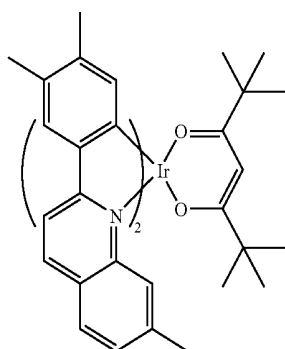
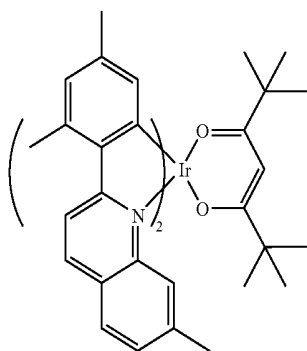
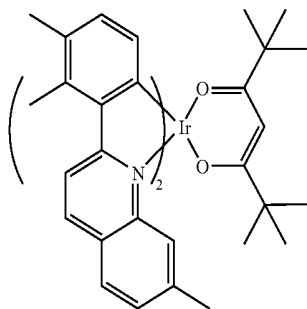
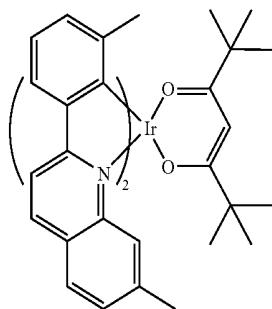
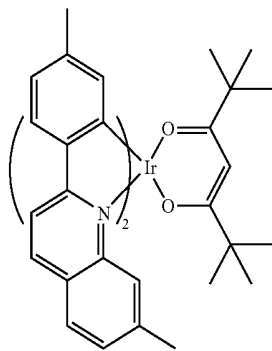
A-11

B-1

B-2

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-continued



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B-3

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B-4

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B-5

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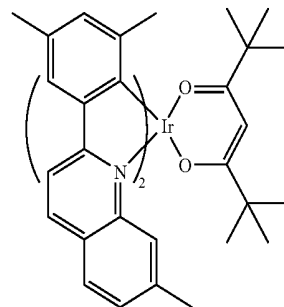
B-6

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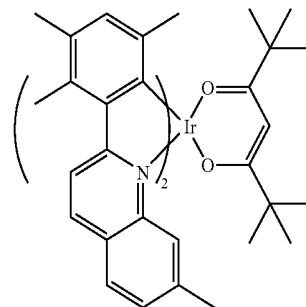
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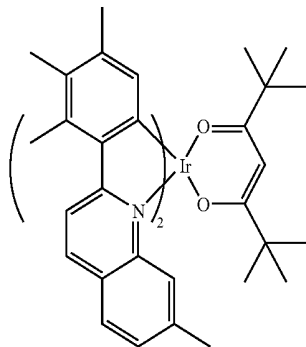
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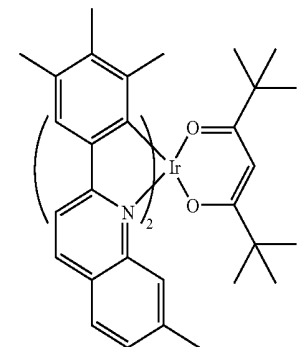
B-9



B-10



B-11



B-7

In another aspect of the present invention, in an organic  
 55 electroluminescence device including an anode, a hole  
 injecting layer, a hole transport layer, a light emitting layer,  
 an electron transport layer, an electron injecting layer, and a  
 cathode serially deposited on one another, the organic elec-  
 60 troluminescence device may use any one of the above-  
 described formulas as a dopant of the light emitting layer.

Herein, any one of Al and Zn metallic complexes and a  
 carbazole derivative may be used as a host of the light  
 emitting layer, and usage of the dopant may be within the  
 range of 0.1 wt. % to 50 wt. %. The efficiency of the present  
 65 invention may be provided when the amount of dopant used  
 is within the above-described range. Furthermore, a ligand  
 of each of the Al and Zn metallic complexes may include



## 11

quinolyl, biphenyl, isoquinolyl, phenyl, methylquinolyl, dimethyl-quinolyl, dimethyl-isoquinolyl, and wherein the carbazole derivative may include CBP.

It is to be understood that both the foregoing general description and the following detailed description of the present invention are exemplary and explanatory and are intended to provide further explanation of the invention as claimed.

## BRIEF DESCRIPTION OF THE DRAWINGS

The accompanying drawings, which are included to provide a further understanding of the invention and are incorporated in and constitute a part of this application, illustrate embodiments of the invention and together with the description serve to explain the principle of the invention. In the drawings:

FIG. 1 illustrates a graph showing a decrease in visibility as color purity of an organic EL device increases (i.e., as the X value of chromacity coordinates becomes greater); and

FIG. 2 illustrates structural formula of NPB, copper (II) phthalocyanine (CuPc), (bpt)<sub>2</sub>Ir(acac), Alq<sub>3</sub>, BAlq, and CBP, which are compounds used in embodiments of the present invention.

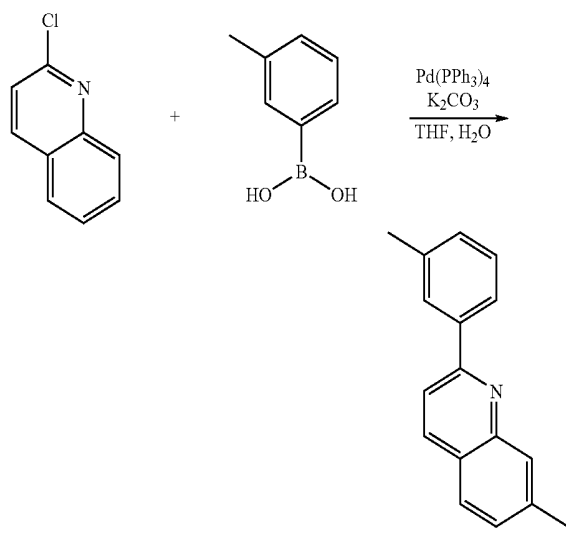
## DETAILED DESCRIPTION OF THE INVENTION

Reference will now be made in detail to the preferred embodiments of the present invention, examples of which are illustrated in the accompanying drawings. Wherever possible, the same reference numbers will be used throughout the drawings to refer to the same or like parts.

A method of combining the red phosphorescence compound according to the present invention will now be described. An iridium (III)(2-(3-methylphenyl)-7-methyl-quinolinato-N,C<sup>2</sup>)(2,4-pentanedionate-0,0) compound, which is shown as A-2 among the red phosphorescence compounds used in the organic EL device according to the present invention.

## COMBINATION EXAMPLE

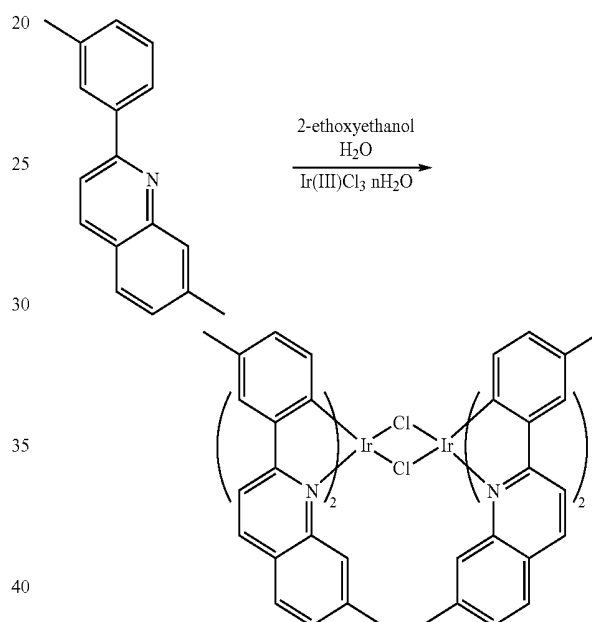
## 1. Combination of 2-(3-methylphenyl)-7-methyl-quinoline



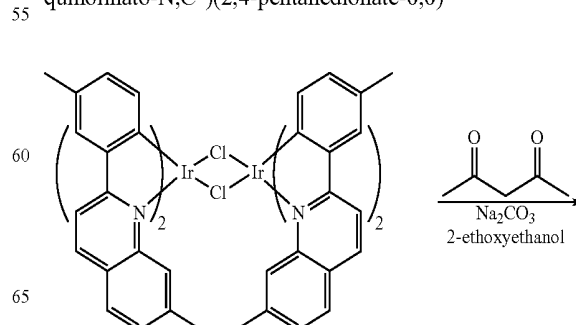
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3-methyl-phenyl-boric acid (1.3 mmol), 2-chloro-7-methyl-quinoline (1 mmol), tetrakis (triphenyl phosphine) palladium(0) (0.05 mmol), and potassium carbonate (3 mmol) are dissolved in a two-neck round bottom flask containing THF (30 ml) and  $\text{H}_2\text{O}$  (10 ml). The mixture is then stirred for 24 hours in a bath of 100 degrees Celsius ( $^{\circ}\text{C}$ ). Subsequently, when reaction no longer occurs, the THF and toluene are discarded. The mixture is extracted by using dichloromethane and water, which is then treated with vacuum distillation. Then, after filtering the mixture with a silica gel column, a solvent is treated with vacuum distillation. Thereafter, by using dichloromethane and petroleum ether, the mixture is re-crystallized and filtered, thereby yielding solid 2-(3-methylphenyl)-7-methyl-quinoline.

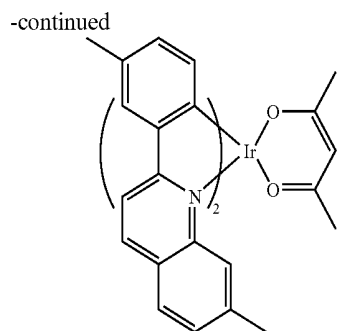
## 2. Formation of Chloro-cross-linked Dimer Complex



Iridium chloride (1 mmol) and 2-(3-methylphenyl)-7-methyl-quinoline (2.5 mmol) are mixed in a 3:1 liquid mixture (30 ml) of 2-ethoxyethanol and distilled water. Then, the mixture is refluxed for 24 hours. Thereafter, water is added so as to filter the solid form that is produced. Subsequently, the solid form is washed by using methanol and petroleum ether, thereby yielding the chloro-cross-linked dimer complex.

3. Formation of Iridium (III)(2-(3-methylphenyl)-7-methyl-quinolinato-N,C<sup>2</sup>)(2,4-pentanedionate-0,0)

13



A chloro-cross-linked dimer complex (1 mmol), 2,4-pentane dione (3 mmol), and  $\text{Na}_2\text{CO}_3$  (6 mmol) are mixed into 2-ethoxyethanol (30 ml) and refluxed for 24 hours. The refluxed mixture is then cooled at room temperature. Thereafter, distilled water is added to the cooled mixture, which is filtered so as to yield a solid form. Subsequently, the solid form is dissolved in dichloromethane, which is then filtered by using silica gel. Afterwards, the dichloromethane is treated with vacuum suction, and the solid form is washed by using methanol and petroleum ether, so as to obtain the chemical compound.

Hereinafter, examples of preferred embodiments will be given to describe the present invention. It will be apparent that the present invention is not limited only to the proposed embodiments.

## EMBODIMENTS

### 1. First Embodiment

An ITO glass substrate is patterned to have a light emitting area of 3 mm×3 mm. Then, the patterned ITO glass substrate is washed. Subsequently, the substrate is mounted on a vacuum chamber. The standard pressure is set to  $1 \times 10^{-6}$  torr. Thereafter, layers of organic matter are formed on the ITO substrate in the order of CuPC (200 Å), NPB (400 Å), CBP+(btp)<sub>2</sub>Ir(acac)(7%) (200 Å), a hole blocking layer (100 Å), Alq<sub>3</sub> (300 Å), LiF (5 Å), and Al (1000 Å).

When forming a hole blocking layer using BAQ, the brightness is equal to 689 cd/m<sup>2</sup> (8.1 V) at 0.9 mA. At this

14

### 2. Second Embodiment

An ITO glass substrate is patterned to have a light emitting area of 3 mm×3 mm. Then, the patterned ITO glass substrate is washed. Subsequently, the substrate is mounted on a vacuum chamber. The standard pressure is set to  $1 \times 10^{-6}$  torr. Thereafter, layers of organic matter are formed on the ITO substrate in the order of CuPC (200 Å), NPB (400 Å), BAQ+A-2(7%) (200 Å), Alq<sub>3</sub> (300 Å), LiF (5 Å), and Al (1000 Å).

At 0.9 mA, the brightness is equal to 1448 cd/m<sup>2</sup> (6.2 V). At this point, CIE x=0.644, y=0.353. Furthermore, the durability (half of the initial brightness) lasts for 8000 hours at 2000 cd/m<sup>2</sup>.

### 3. Third Embodiment

An ITO glass substrate is patterned to have a light emitting area of 3 mm×3 mm. Then, the patterned ITO glass substrate is washed. Subsequently, the substrate is mounted on a vacuum chamber. The standard pressure is set to  $1 \times 10^{-6}$  torr. Thereafter, layers of organic matter are formed on the ITO substrate in the order of CuPC (200 Å), NPB (400 Å), BAQ+A-5(7%) (200 Å), Alq<sub>3</sub> (300 Å), LiF (5 Å), and Al (1000 Å).

At 0.9 mA, the brightness is equal to 1378 cd/m<sup>2</sup> (6.0 V). At this point, CIE x=0.659, y=0.351. Furthermore, the durability (half of the initial brightness) lasts for 7000 hours at 2000 cd/m<sup>2</sup>.

### COMPARISON EXAMPLE

An ITO glass substrate is patterned to have a light emitting area of 3 mm×3 mm. Then, the patterned ITO glass substrate is washed. Subsequently, the substrate is mounted on a vacuum chamber. The standard pressure is set to  $1 \times 10^{-6}$  torr. Thereafter, layers of organic matter are formed on the ITO substrate in the order of CuPC (200 Å), NPB (400 Å), BAQ+(btp)<sub>2</sub>Ir(acac) (7%) (200 Å), Alq<sub>3</sub> (300 Å), LiF (5 Å), and Al (1000 Å).

At 0.9 mA, the brightness is equal to 780 cd/m<sup>2</sup> (7.5 V). At this point, CIE x=0.659, y=0.329. Furthermore, the durability (half of the initial brightness) lasts for 2500 hours at 2000 cd/m<sup>2</sup>.

In accordance with the above-described embodiments and comparison example, the characteristics of efficiency, chromaticity coordinates, brightness, and durability are shown in Table 1 below.

TABLE 1

Device	Voltage (V)	Current (mA)	Brightness (cd/m <sup>2</sup> )	Current Efficiency (cd/A)	Power Efficiency (1 m/W)	CIE (X)	CIE (Y)	Durability(h) ½ of initial brightness
First Embodiment	8.1	0.9	690	6.9	2.7	0.651	0.329	1600
Second Embodiment	6.2	0.9	1450	14.5	7.3	0.644	0.353	8000
Third Embodiment	6.0	0.9	1378	13.8	7.2	0.659	0.351	7000
Comparison Example	7.5	0.9	780	7.8	3.3	0.659	0.329	2500

point, CIE x=0.651, y=0.329. Furthermore, the durability (half of the initial brightness) lasts for 1600 hours at 2000 cd/m<sup>2</sup>.

As shown in Table 1, the device is operated with high efficiency at a low voltage even when the color purity is high. Furthermore, the current efficiency of the second embodiment has increased by 100% or more as compared to

15

the comparison example. Additionally, the durability of the second embodiment has increased to three times that of the comparison example.

Table 2 below shows the characteristics of efficiency, chromacity coordinates, and brightness in accordance with the increase in voltage and electric current in the organic electroluminescence device according to the second embodiment of the present invention.

TABLE 2

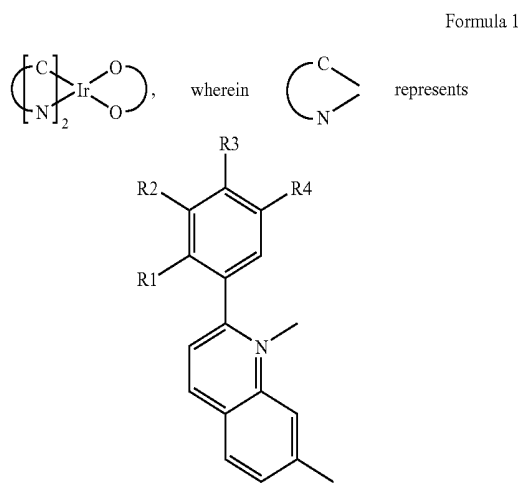
Voltage (V)	Current (A/(mA/cm <sup>2</sup> ))	Brightness (cd/m <sup>2</sup> )	Current Efficiency (cd/A)	Power Efficiency (1 m/W)	CIE (X)	CIE (Y)
5.0	1.111	168.6	15.2	9.5	0.645	0.353
5.5	3.333	500.8	15.0	8.5	0.645	0.353
6.0	7.777	1139	14.6	7.6	0.644	0.354
6.5	16.666	2309	13.9	6.6	0.643	0.354
7.0	33.333	4275	12.9	5.7	0.643	0.355
7.5	66.666	7664	11.5	4.8	0.641	0.356

As shown in Table 2, the second embodiment provides excellent efficiency, and the chromacity coordinates according to the driving voltage also maintains high color purity.

It will be apparent to those skilled in the art that various modifications and variations can be made in the present invention without departing from the spirit or scope of the inventions. Thus, it is intended that the present invention covers the modifications and variations of this invention provided they come within the scope of the appended claims and their equivalents.

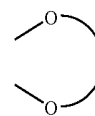
What is claimed is:

1. An organic electroluminescent device comprising a light emitting layer, the light emitting layer comprising a red phosphorescence compound of Formula 1 below as a dopant:



wherein each of R1, R2, R3, and R4 is independently selected from the group consisting of hydrogen and substituted or non-substituted C1 to C6 alkyl groups, and wherein

16



represents an alkanedione selected from the group consisting of 2,4-pentanedione, 2,2,6,6-tetra-methylheptane-3,5-dione, 3,5-heptanedione, 1,1,1-trifluoro-2,4-pentanedione, 1,1,1,5,5,5-hexafluoro-2,4-pentanedione, and 2,2-dimethyl-3,5-hexanedione.

2. The organic electroluminescent device of claim 1, wherein any one of Al and Zn metallic complexes and carbazole derivatives is used as a host of the light emitting layer.

3. The organic electroluminescent device of claim 1, wherein any one of Al and Zn metallic complexes and a carbazole derivative is used as a host of the light emitting layer, and wherein usage of the dopant within the range of 0.1 wt. % to 50 wt. %.

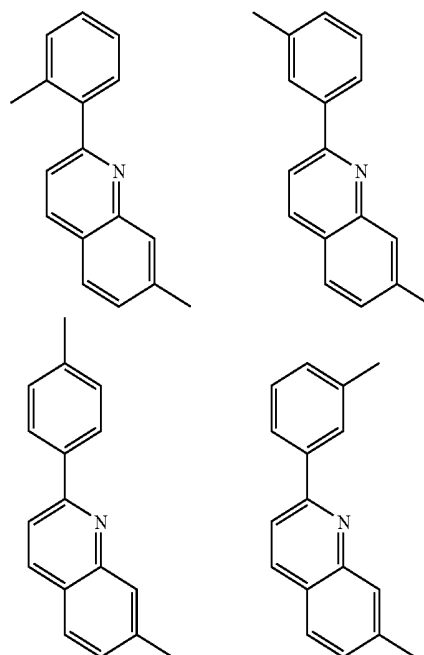
4. The organic electroluminescent device of claim 2, wherein a ligand in each of the Al and Zn metallic complexes comprises quinolyl, biphenyl, isoquinolyl, phenyl, methylquinolyl, dimethylquinolyl, dimethyl-isoquinolyl, and wherein the carbazole derivative comprises CBP.

5. The organic electroluminescent device of claim 1, wherein each of the C1 to C6 alkyl groups is selected from a group consisting of methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, and t-butyl.

6. The organic electroluminescent device of claim 1, wherein



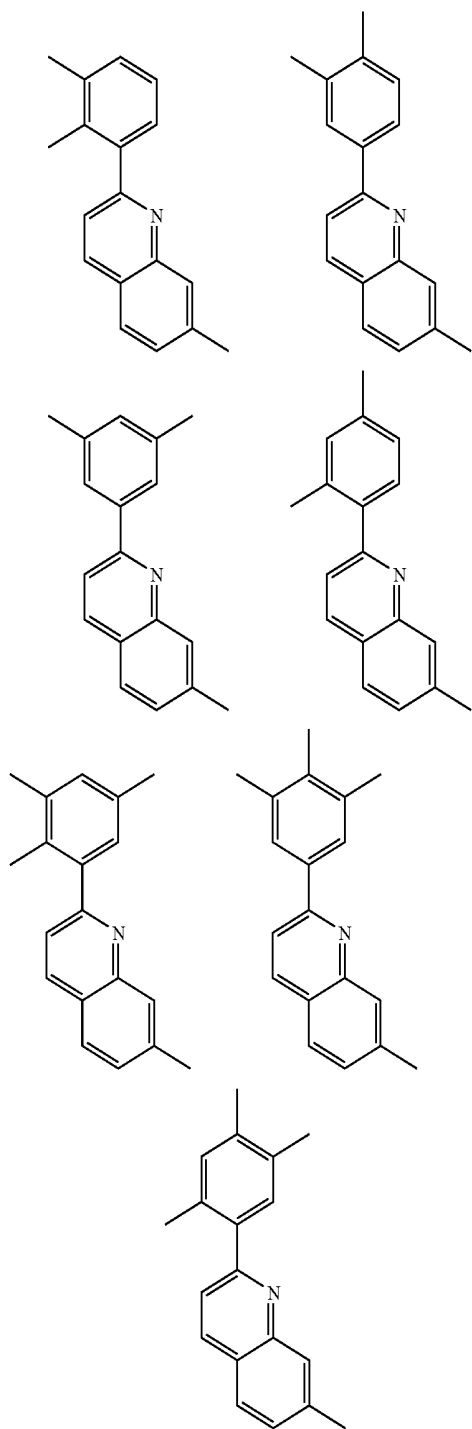
is any one of the following chemical formulae:



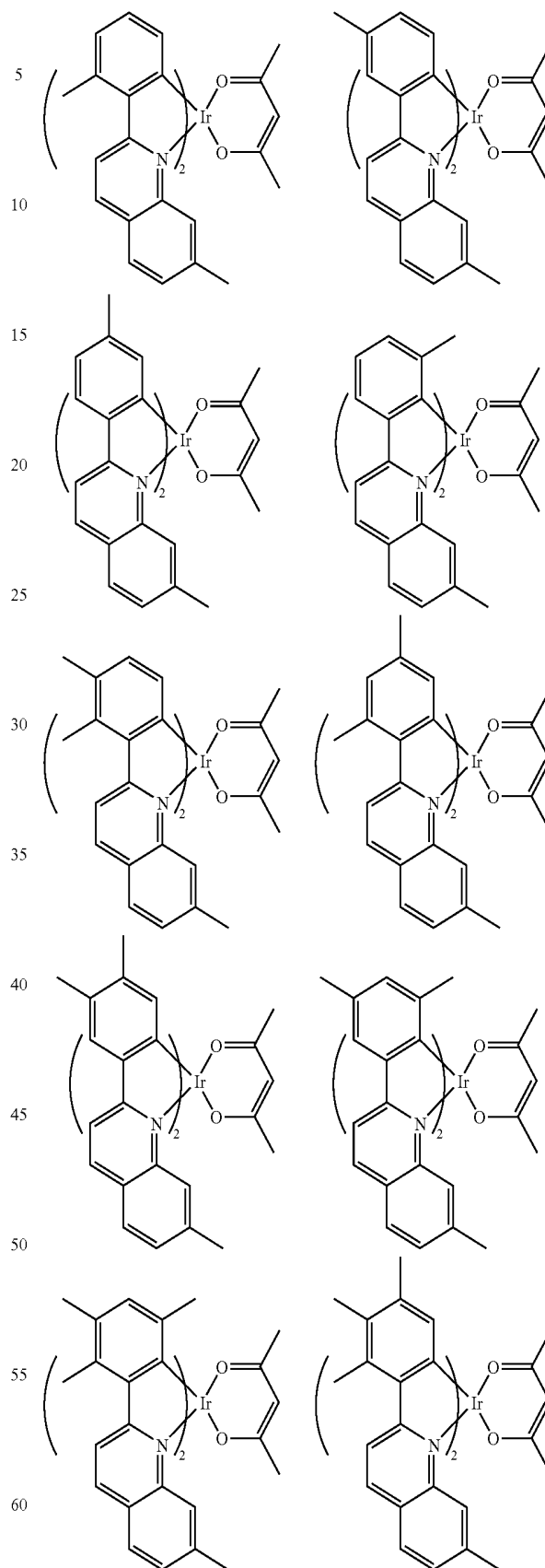
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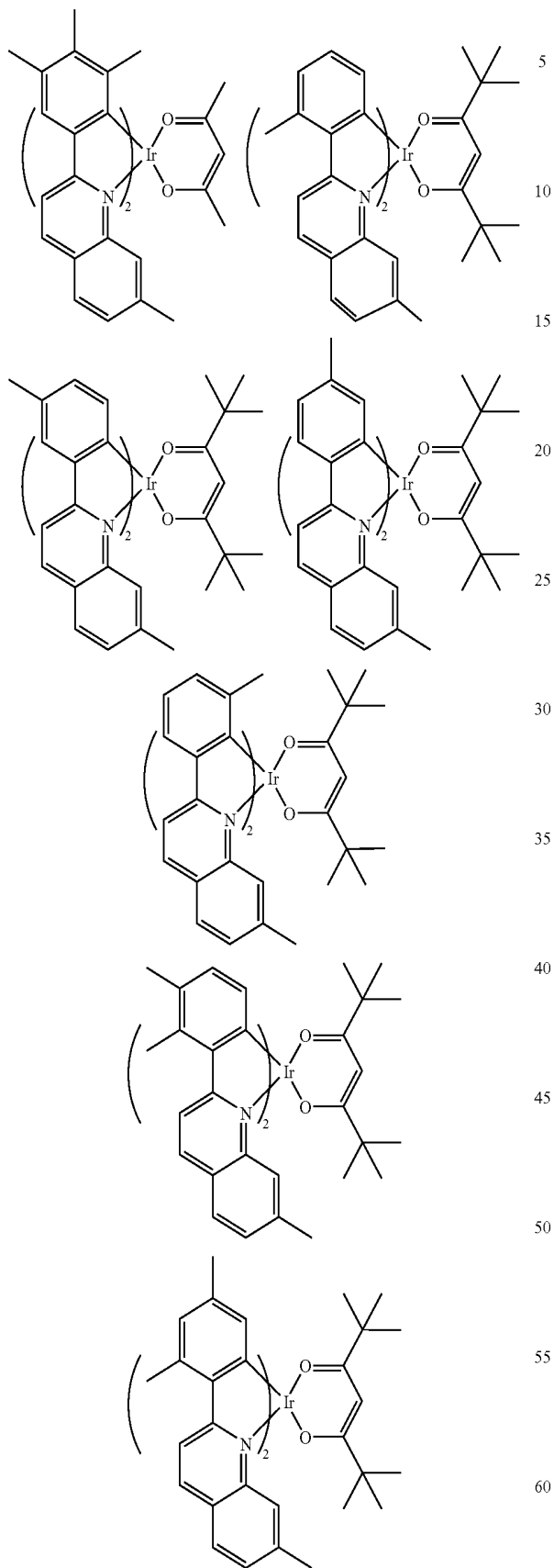
and



7. The organic electroluminescent device of claim 1, wherein the Formula 1 is any one of the following chemical formulae:

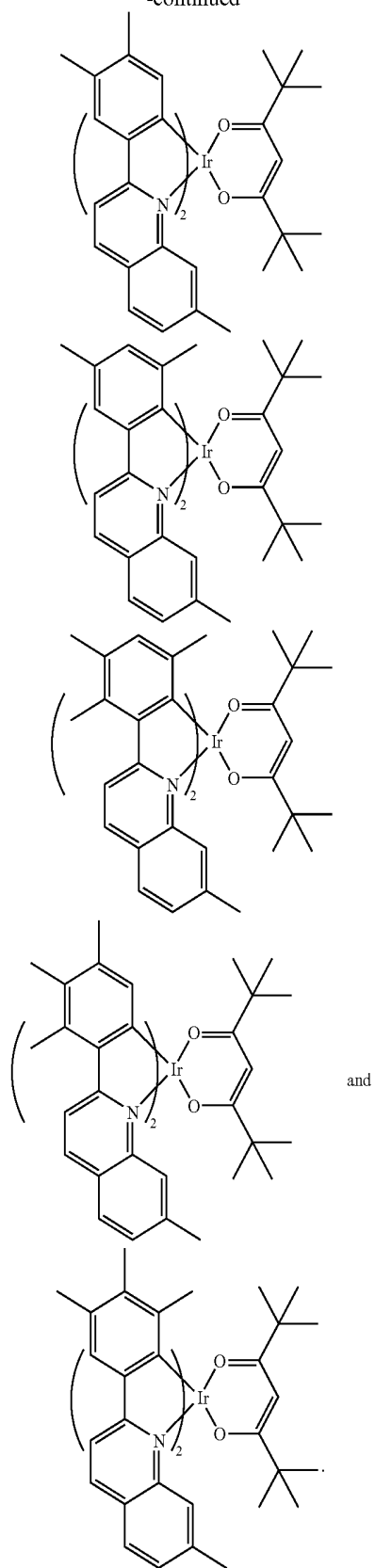
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专利名称(译)	使用红色磷光化合物的有机电致发光器件		
公开(公告)号	<a href="#">US7378162</a>	公开(公告)日	2008-05-27
申请号	US11/128421	申请日	2005-05-13
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IPC分类号	H05B33/14 C07D215/00 C09K11/06		
CPC分类号	C07F15/0033 C09K11/06 H01L51/0085 H05B33/14 C09K2211/1007 C09K2211/1029 C09K2211/1092 C09K2211/185 C09K2211/186 C09K2211/188 H01L51/0052 H01L51/0059 H01L51/0062 H01L51/0078 H01L51/0081 H01L51/5016 Y10S428/917		
优先权	1020050019182 2005-03-08 KR		
其他公开文献	US20060202194A1		
外部链接	<a href="#">Espacenet</a> <a href="#">USPTO</a>		

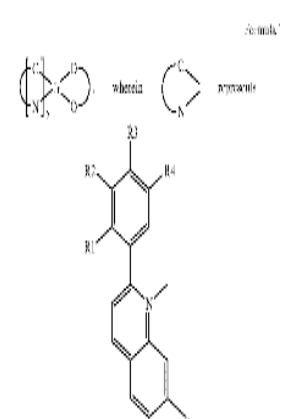
#### 摘要(译)

公开了红色磷光化合物和使用其的有机电致发光器件。在包括阳极，空穴注入层，空穴传输层，发光层，电子传输层，电子注入层和阴极彼此串联沉积的有机电致发光器件中，有机电致发光器件可以使用化合物作为发光层的掺杂剂。

(30) Foreign Application Priority Data  
 Mar. 8, 2005 (KR) 10-2005-0019182

(51) Int. Cl.  
*H05B 33/14* (2006.01)  
*C09K 11/06* (2006.01)  
*C07D 215/00* (2006.01)  
 (52) U.S. Cl. 428/690, 428/917, 257/40;  
 313/504; 546/4; 546/10  
 (58) Field of Classification Search 313/504;  
 546/4, 2, 10, 257/40, 428/690, 917  
 See application file for complete search history.

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7 Claims, 2 Drawing Sheets